## CHM 6680

## Characterization of Paramagnetic Molecules Section: 18844 Credits: 3

Instructor: Prof. George Christou Office: CLB 408A Email: <u>christou@chem.ufl.edu</u>

Class Times: T, R: Periods 3-4 (Class will be 9.45 – 11.10 am) Room: LEI 242

- Office Hours: R 2.00 4.00 pm, and by appointment
- **Description:** Metal-containing paramagnetic molecules are commonly encountered in chemistry and biochemistry, in fields ranging from inorganic chemistry to the active sites of metalloproteins and metalloenzymes. In all these areas, the presence of the unpaired electron(s) has a major impact on the observed physical and spectroscopic properties. Many chemical and biochemical systems contain mononuclear metal ions, whereas others contain polynuclear sites comprising two or more interacting metal ions. The course will therefore cover the qualitative and quantitative tools and theories needed to understand the magnetic and spectroscopic properties of molecular metal compounds (both mononuclear and multinuclear) possessing unpaired electrons.
- Outline: Basic concepts and definitions in magnetism; Magnetization and magnetic susceptibility; Spin and orbital angular momentum; Van Vleck equation; Transition metals vs lanthanides; Zero-field splitting in mononuclear metal compounds; High-spin/low-spin equilibrium, and spin crossover; Polynuclear compounds and intra-molecular exchange interactions; Mechanisms of antiferromagnetic and ferromagnetic interactions; Spin frustration in polynuclear compounds; Single-molecule magnets (molecular nanomagnets); NMR of paramagnetic molecules; EPR of mononuclear paramagnetic metal complexes (time permitting).
- **Grade:** The course grade will be determined by two exams during the semester (30% each) and a Final Exam (40%).
- Final Exam: 5.30 7:30 pm on May 2, 2024

## **Reference Texts (no required text):**

- 1. Molecular Magnetism, O. Kahn, VCH Publishers, Mannheim, 1993
- 2. C. J. O'Connor, Prog. Inorg. Chem. 1982, 29, 203
- 3. *Physical Methods in Chemistry*, R. S. Drago, 1<sup>st</sup> or 2<sup>nd</sup> Edition
- 4. Magnetochemistry, R. L. Carlin, Springer-Verlag, Berlin, 1986
- 5. *Magnetism and Transition Metal Complexes,* F.E. Mabbs and D.J. Machin, Dover Publications, New York, 2008
- 1. Other Information: Student Honor and Conduct Codes: see https://sccr.dso.ufl.edu/policies/student-honor-code-student-conduct-code/
- 2. Students with Disabilities: see https://disability.ufl.edu/
- 3. Counseling and Health Care: see https://counseling.ufl.edu/