Syllabus

CHM 3120L ANALYTICAL CHEMISTRY LABORATORY

Fall 2022

Faculty Instructor: Dr. Alexander Jacobs, Leigh 202A

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Office Hours: Friday 1pm-2pm Leigh Hall 202

Simultaneously by Zoom ID: 882 880 7636 Password: Officehr

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Course Website: Canvas; Please visit the website regularly for announcements and resources. Everything is posted under "Files"

Videos available at: https://www.chem.ufl.edu/undergraduate/courses-and-curriculum/chemistry-

Required Materials

Laboratory Manual: No lab manual is required. All materials will be posted on the e-learning site, under Files.

Laboratory Notebook: Any sensible laboratory notebook, to be used only for this lab, is suitable. You will turn in scans or

photos of your notes, retaining the original notebook for your own use. Please be sure that what you

submit is legible and clear.

Laboratory Attire: The Essentials: Long, loose-fitting pants, full shirt, shoes which cover the feet, departmentally-

approved safety glasses, tie-back for long hair.

Course Objectives

CHM 3120L is an introductory laboratory course in Analytical Chemistry. By the end of the semester, students are expected to demonstrate:

- proper laboratory techniques for quantitative chemical measurements including accuracy on unknowns
- knowledge of a select group of analytical methods
- · competence in data analysis, statistics and preparation of professional laboratory reports

Grading

Your grade will be determined by the accuracy of your results, the quality of your reports, the quality of your laboratory notes, your competence in essential laboratory manipulations, and your performance on written quizzes. Grades will be posted in the Canvas gradebook.

Accuracy	6 @ 35 points	210
Questions, Reports and Notes	8 @ 70 points	560
Practical Exams	4 @ 40 points	160
Written Quizzes	3 @ 45 points _	135
		1065_total

The following grading scale will be used:

Letter Gra	de Percentage	Letter G	rade Percentage	Letter Grade	Percentage
A	≥93	B-	≥80	D+	≥67
A-	≥90	C+	≥77	D	≥64
B+	≥87	C	≥73	E	<60
В	≥83	C-	≥70		

There may be a curve, there may not be a curve.

Notes:

- 1) Prior to the first lab, visit the e-learning site and review Preliminary Handouts 1-3, 5: laboratory safety, basic lab rules, laboratory notebook, and fundamental techniques. Also read the handout for Experiment #1.
- 2) You will need to bring a copy of the lab experiments to your lab period. This can be done with either a print out, on a tablet or a laptop. It is advised not to use your phone.
- 3) On the first day, you select a workstation. This is where you will complete your experiments. Everything you will need will be at that station.
- 4) A minimum of 18 out of 35 accuracy points will be given if the experiment is performed, the results are calculated correctly, and deadlines are met. For labs with an unknown number, the number must be reported or will get an accuracy score of zero.
- 5) For labs 1-5, there will be questions at the end of the lab handout. You will turn in the answers to these questions and calculations along with your results, tabulated and presented nicely. Starting with lab 2, you will be asked to write a section of a lab report (Intro, Experimental, Results and Discussion and Conclusion) with each lab. For lab 6 and 7, you will be writing a full lab report. More details will be given about each section as the semester goes on. There are some examples of lab reports on Canvas. Videos covering scientific writing will be posted to Canvas.
- 6) Lab reports and answers to questions must be typed. By lab 2, all calculations must be shown through Microsoft Equation Editor.
- 7) A copy of your lab notes must be submitted with your reports.
- 8) Lab reports are due at the specified time on Canvas for your section
- 9) A 10% penalty off the final score of the report will be assessed each time a result or report is submitted late. <u>The</u> maximum permissible late time is one week.
- 10) All written work (late or otherwise) must be received by 3:00 PM on Friday, 12/7/22.
- 11) Each student is expected to pass laboratory practical exams on four essential analytical skills (use of the analytical balance/weighing by difference, quantitative transfer/use of a volumetric flask, use of glass pipets and use of

micropipettes). The tests will be given by the TA during the regular laboratory period at times mutually acceptable to both the student and the TA.

- 12) Three written quizzes will be given on the dates specified on the schedule. Quizzes will be given on Canvas. They are open note, but you must work alone. If you wish to go over your quiz, please contact your TA. The questions in your lab write ups will greatly help you prepare for the quizzes.
- 13) Attendance is required at all scheduled laboratory periods, unless you are informed otherwise by your TA or the instructor.
- 14) Plagiarism will not be tolerated. Students are expected to obey the University of Florida Honor Code, detailed at

The Honor Code (http://www.dso.ufl.edu/sccr/process/student-conduct-honor-code/) specifies a number of behaviors that are in violation of this code and the possible sanctions. Furthermore, you are obligated to report any condition that facilitates academic misconduct to appropriate personnel. If you have any questions or concerns, please consult with the instructors or TAs in this class.

You are required to abide by the Student Honor Code. Any violation of the academic integrity expected of you will result in a minimum academic sanction of a zero on the assignment. Any alleged violations of the Student Honor Code will result in a referral to Student Conduct and Conflict Resolution. Please review the Student Honor Code and Student Conduct Code at https://sccr.dso.ufl.edu/policies/student-honor-code-student-conduct-code

- 15) Make-ups will be granted only when justified. If you know ahead that you will have to miss lab, notify your TA and Dr. Jacobs in advance. If you are sick and cannot reach anyone before lab, you will have to present written evidence of the illness. *If you are not feeling well, do not come to lab*.
- 16) If you are involved in a laboratory accident, you <u>must</u> go to the infirmary for treatment.
- 17) Students requesting classroom accommodation must first register with the Dean of Students Office. The Dean of Students Office will provide documentation to the student who must then provide this documentation to the instructor when requesting accommodation.

Course Fees: You have already paid this, but the fee for this course is \$45.00

Laboratory Schedule
Note: Note that the schedule may change due to weather/holidays/pandemic

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Dates (starting	Preparation	Lab Work	Quizzes	Assignment Due
date)				
Begin Aug 23 rd		No lab (add/drop week)		
Week 2 (week of	Read Handouts 1-6	Check in		
Aug 29)	Read Experiment 1	Experiment 1		
	•	Balance use		
		Pipet use/calibration		
Week 3 (week of		No lab due to Labor Day		
Sept 5 th)				
Week 4, (week of	Read Handout 7	Begin Soda Ash Titrations		Experiment 1 results and lab
Sept 12 th)	Read Experiment 2	HCl/NaOH titrations		notes
Sept 12)	Troub Emperation 2	11021140114242515		115005
Week 5 (week of	Review handouts to	KHP/NaOH titrations	Quiz 1 and	
Sept 19 th)	prepare for Quiz 1	Finish Soda Ash	Deadline for	
Sept 19	prepare for Quiz 1	I mish sodd 7 ish	Weighing	
			Practical	
Weels 6 (weels of	Dood Evmoniment 2	Beer's Law measurement	Tractical	Code Ash questions mosults
Week 6 (week of	Read Experiment 3	Beer's Law measurement		Soda Ash questions, results, Introduction and notes due
Sept 26 th) Week 7 (week of		No Class due to Hamasania		introduction and notes due
`		No Class due to Homecoming		
Oct 3 rd)				
Week 8 (week of	Read Experiment 4	Spectrophotometric Fe		Beer's Law questions, results,
Oct 10 th)				Experimental and notes due
Week 9, (week of		Spectrophotometric Fe	Quiz 2	
Oct 17 th)		Finish Lab 4	Deadline for	
			Pipetting Practical	
Week 10 (week of	Read Experiment 5	Chloride Ion Selective		Spec Fe questions, results,
Oct 24 th)	Watch ISE video	Electrodes		Results and Discussion and
				notes due
Week 11 (week of	Read Experiment 6	Fluorescence of Quinine;	Deadline for	ISE questions, results,
Oct 31st)	1	Standard additions and	volumetric flask	Conclusion and lab notes due
,		Determination of quinine in	practical	
		cinchona bark	1	
Week 12 (week of	No in person lab	At home Chromatography		Quinine in tonic water report
Nov 7 th)	this week	assignment		and notes due (Friday class
				due on Monday Nov 14th)
Week 13 (week of	Read Experiment 7	Chromatography of Soda	Quiz 3	At home Chromatography
Nov 14 th)	Troug Experiment /	Cinomatography of boda	Deadline for	assignment
			Micropipette	
			Practical	
Week 14 (week of		No lab due to Thanksgiving	1 Iucticui	
Nov 21st)		110 lab due to Thanksgiving		
Week 15 (week of				Chromatography of Soda
Nov 28 th)				report and notes due
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