

## General Chemistry II CHM 2046 (Lopez, online, Spring 2021)

*The instructor reserves the right to make changes or corrections to this syllabus at any time.  
Students will be notified when any changes are made via an announcement on canvas.*

### Course Overview

**DESCRIPTION:** CHM 2046 and CHM 2046L constitute the second semester of the two-term sequence of General Chemistry, CHM 2045/2045L - 2046/2046L. Prerequisite information and credit suitability can be found in the Undergraduate Catalog. Requirements for class attendance and make-up exams, assignments, and other work in this course are consistent with university policies that can be found at: <https://catalog.ufl.edu/ugrad/current/regulations/info/attendance.aspx>

**COURSE OBJECTIVES:** As both a general education requirement and majors course, CHM2046 serves to teach: the scientific method, skills for problem solving, general chemistry knowledge, and a connection to the principles that govern the natural world.

**LECTURE:** Online, asynchronous (suggested days to watch Lectures: M, W, F),

<p><b>Lectures:</b> January 11th through April 21st <b>Instructor: Dr. Simon E. Lopez</b> <b>Office:</b> LEI #312 (located at Leigh Hall). Office hours will be online using “zoom”. <b>E-mail</b> (for administrative purposes only): <a href="mailto:simonlopez@chem.ufl.edu">simonlopez@chem.ufl.edu</a></p>
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<p style="text-align: center;"><u>Office Hours (zoom)</u> M,W,F 11:30 am - 1:00 pm</p>
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### MATERIALS:

**Recommended:** Chemistry Textbook by Silberberg (8<sup>th</sup> edition) + Other College level General Chemistry Books. There is an option to opt-in to the Silberberg 8<sup>th</sup> edition eText for a discounted price via UF All Access for a limited time. (copies of the 8<sup>th</sup> ed and solution manual are available in the Marston Science library).

**Optional:** eBook of Student Solutions Manual made specifically for this course. Title: CHM 2046 – SSM ISBN: 9781307629750 and can be purchased here: <https://create.mheducation.com/shop/> through McGraw Hill or through UF Bookstore.

**Required: Non-graphic / non-programmable scientific** calculator. You may **not** use your phone as a calculator.

**Technology Requirement:** To ensure the most effective learning environment, it is important that you are equipped with the following: reliable internet connection, regular access to a laptop or desktop computer with an operating system and web browser (specifically Google Chrome for taking proctored exams using HonorLock), and also webcam, microphone, and speakers for virtual class sessions and proctored exams.

**Note:** The University of Florida provides various software licenses to students free of charge or at discounted rates. Please visit <https://software.ufl.edu/software-listings/> to access products that may interest you, for example Microsoft Office 365 ProPlus.

**GRADES:** Grades for the term will be determined as follows:

3 Progress Exams	60%
Final Cumulative Exam	23%
Online Homework	10%
Worksheets	2%
Weekly Quizzes	5%
TOTAL	100%

The following grade cutoffs will be used (these are non-negotiable):

92-100% = A	88-91.9% = A-	84-87.9% = B+	80-83.9% = B	76-79.9% = B-
72-75.9% = C+	68-71.9% = C	64-67.9% = D+	60-63.9% = D	56-59.9% = D-
< 56.0% = E				

Information on current UF grading policies for assigning grade points can be found at:

<https://catalog.ufl.edu/ugrad/current/regulations/info/grades.aspx>

**POSTED GRADES:** Should a student wish to dispute any grade received in this class, the dispute must be in writing (via e-mail to [simonlopez@chem.ufl.edu](mailto:simonlopez@chem.ufl.edu)) and submitted to the instructor within one week of the grade being posted to canvas. After one week has passed from when the grade was posted and the student made aware of the posting of the grade(s) via an announcement on canvas, the instructor considers those grades final.

**ONLINE HOMEWORK:** Ten percent of the course grade (10 %) will be based on online homework assignments through the Canvas website of CHM2046. Each assignment has a displayed deadline for earning full credit; assignments that are late can be completed for half credit (maximum 24 h later, contact Dr. Lopez). Students that miss a homework deadline due to an excused absence can ask for an extension by contacting the instructor. You can earn up to 10 % toward your grade by completing these assignments.

**WEEKLY POST-LECTURE QUIZZES (PLQ):** After the Drop/Add period ends, you will have two (2) weekly quizzes related to content previously covered in lectures before the due date for each quiz. Quizzes will contain 5-6 questions each related to lecture content. Questions can be structured in multiple formats (*i.e.* multiple-choice, multiple-dropdowns, fill in the blanks, numeric entries, etc) and must be answered in a total time of 30 minutes. PLQs will be due every Tuesday & Thursday (11:59 pm). These quizzes will be available every Monday morning from Canvas, you should carefully read the instructions for every quiz before you start. Check periodically (daily) your canvas site of the course, emails and announcements. For any missed quiz, check the policy applied for missed exams described below in this syllabus (Exam Absences). Same policy will apply in this case.

**CONTACTING THE INSTRUCTOR / OFFICE HOURS:** Emails are for administrative purposes only, and not for distance-instruction. All academic inquiries must be made during office hours (zoom meeting) or before/after lectures (by previous appointment by zoom meeting). Please be prepared before coming to office hours, bring specific questions and your previous work

**University of Florida**  
**CHM 2046 General Chemistry 2**  
**Spring 2021; Course Schedule**

**Instructor:** Dr. Simon E. Lopez, 312 Leigh Hall (Office).

**E-mail:** [simonlopez@chem.ufl.edu](mailto:simonlopez@chem.ufl.edu) (for administrative purposes); Phone: (352)392-9700.

**E-Learning (Canvas):** <https://lss.at.ufl.edu/> (will be updated regularly).

**Classes: Online, asynchronous (suggested days to watch lectures: M,W,F).**

**Location: online- Canvas**

**Office Hours:** M, W, F (11:30 a.m. – 1:00 p.m.), Zoom.

Week	Dates (suggested to watch lectures)	Topics (# of Lectures)	Chapters*
Jan 11 <sup>th</sup>	Jan 11,13,15	Review of Chemical Kinetics; Reaction Rates; Activation Energy; Reaction Mechanisms; Catalysts (3)	Ch. 16
Jan 18 <sup>th</sup>	Jan 19,20,22	Gas-phase and Heterogeneous Equilibria; Le-Chatelier's Principle (3)	Ch. 17
Jan 25 <sup>th</sup>	Jan 25, 27 Jan 29	Reaction yield: Optimization (2) Brønsted-Lowry acid-base potential and equilibria (1)	Ch 17 Ch. 18.1-18.8
Feb 1 <sup>st</sup>	Feb 1,3,5	Brønsted-Lowry acid-base potential and equilibria (2) Lewis Acid-Base, Electrophile-Nucleophile, reactions and mechanisms (1)	Ch. 18.1-18.8 Ch. 18.9
<b>Feb 9<sup>th</sup> (Tuesday): Progress Exam 1 (8:20-10:20 pm); Cumulative Ch. 16-18</b>			
Feb 8 <sup>th</sup>	Feb 8, 10, 12 + Feb 15, 17	Acid-Base Titrations and Buffers (5)	Ch. 19.1-19.2
Feb 15 <sup>th</sup>	Feb 19	Solubility Equilibria (1)	Ch. 19.3
Feb 22 <sup>nd</sup>	Feb 22,24,26	Solubility Equilibria, selective precipitation (3)	Ch. 19.3
March 1 <sup>st</sup>	March 1,3 March 5	Complex ions and Amphoteric Hydroxides (2) Equilibrium Thermodynamics (1)	Ch. 19.4 Ch. 20
<b>March 10<sup>th</sup> (Wednesday): Progress Exam 2 (8:20-10:20 pm); Cumulative Ch. 16-19.4</b>			
March 8 <sup>th</sup>	March 8,10,12 + March 15	Equilibrium Thermodynamics (4)	Ch. 20
March 15 <sup>th</sup>	March 17,19	Oxidation/Reduction Potential: Redox Reactions (2)	Ch. 21
March 22 <sup>nd</sup>	March 22,24,26 + March 29	Electrochemistry and Equilibria; Voltaic & Electrolytic Cells (4)	Ch. 21
March 29 <sup>th</sup>	March 31, April 2 + April 5	Transition Metals (3)	Ch 23
<b>April 7<sup>th</sup> (Wednesday): Progress Exam 3 (8:20-10:20 pm); Cumulative Ch. 16-23</b>			
April 5 <sup>th</sup>	April 7 April 9	Review for Exam 3 (1) Introduction to Nuclear Chemistry (1)	Review for Exam 3 Ch. 24
April 12 <sup>th</sup>	April 12,14 April 16	Introduction to Nuclear Chemistry (2) Introduction to Organic Chemistry (1)	Ch. 24 Ch. 15
April 19 <sup>th</sup>	April 19,21	Introduction to Organic Chemistry (2)	Ch. 15
<b>April 24<sup>th</sup> (Saturday): Final Exam (10:00 am-12:00 pm); Cumulative Ch. 15-21+23,24</b>			
<i>Holidays (no classes): January 18<sup>th</sup> (Dr. Martin Luther King Jr. Day)</i>			

**DISCUSSION CLASSES + WORKSHEETS (WS):** The Discussion Classes meet every Thursday and your attendance is expected (attendance will be checked by your TA during the zoom session). The time of your discussion section is synchronous to the posted schedule, as it is the time you must be present. Your discussion section will contain weekly worksheets that will count toward your overall grade (2%). You must go to your assigned discussion section to receive credit for the worksheet. Groups of approximately 3-4 students will be assigned by your TA to work on it together. Any grade discrepancy needs to be addressed within a week of grades posting to canvas. Discussion sessions will be held online using Zoom under the guidance of your graduate TA. You must meet during your scheduled discussion session with your TA who will coordinate the session. Attendance will be taken (2 pts) as well as participation (3 pts). To receive credit for your WS (5 pts) you must attend the zoom meeting and upload your completed WS before 11:59 pm the same day of your discussion. Your attendance and participation will be recorded during the discussions. If you are more than 5 minutes late, then you forfeit your participation points for the day. If you are not present during the first 25 minutes of discussion period, then you forfeit your attendance points for the day. To account for technical issues, one day of attendance/participation points will be dropped for all students. All further absences will be marked as a 0.

**Audio/Video Presence Policy:** For discussion sections, the participation portion of your grade for this class will be calculated on the basis of your attendance and your participation in class activities. Since the pedagogical approach of this course depends heavily on student engagement and interaction, you are required, at a minimum, to participate in class activities through the audio function of Zoom. Your video presence is invited as well. As in all courses, unauthorized recording and unauthorized sharing of recorded materials are prohibited. Full audio and video presence are required for proctored tests administered by Honorlock.

**CHEMISTRY LEARNING CENTER (CLC):** There is free help to be had from graduate student teaching assistants Monday through Friday via office hours in Zoom: the TAs schedule will be announced through Canvas. Your discussion TA will have office hours on Zoom, but you may go any TA's scheduled office hours (listed in Canvas) to get help on questions pertaining to chemistry. Additionally, there is the teaching center located on the ground floor of Broward Hall (they use to offer periodical reviews previous each progress exam and the final exam). Its web site is <http://www.teachingcenter.ufl.edu>.

**CANVAS** (<http://elearning.ufl.edu>): Here you will find the syllabus, gradebook, files, class announcements, and other pertinent info for the course. It is your responsibility to check Canvas often to make sure that you do not miss important announcements and to ensure that your gradebook is accurate. For computer assistance, visit <http://helpdesk.ufl.edu/>.

**CLASS DEMEANOR:** In order to have an optimal learning environment, the classroom needs to be free of disruptions. Therefore, it is expected that students come to class on time and leave only when class is concluded by the instructor, and that the class is not disrupted by student talking or cell phone noises (in case of zoom meetings).

**EXAMS:** Exams will be administered at night from 8:20pm to 10:20pm (Eastern time) outside of class, and will be held through Canvas using Honorlock. Exam questions will consist of questions similar to HWs /worksheets/ practice problems you have completed and practice on Canvas, as well as PLQs questions. So expect numeric entry, multiple dropdowns, multiple answer, true/false, multiple choice, matching, and multiple fill in the blanks. You must use a non-graphing non-programmable scientific calculator on exams (with log, ln, root, and exponent (scientific notation) functions). Be sure to also bring pencils, and your UFID card. No notes, papers, cell phones or other electronic devices can be in view during exams. **Detailed instructions for your exams using Canvas will be given prior the exam.**

This course uses **Honorlock** for proctoring of during-term exams. Honorlock is UF's designated online proctoring service for classroom exams and quizzes that were previously in person but have moved online

as part of the COVID-19 response effort. In order for you to take exams in this course you will need a government issued photo ID (or your Gator-1 ID), a working camera and microphone on your computer, a stable internet connection, and the Google Chrome browser (<https://chrome.com>) on your computer. Before and during your exam you will need to follow the Honorlock proctor's instructions. Please familiarize yourself with the Honorlock student's preparation guide: [Honorlock-Student-Exam-Preparation-Information.pdf \(ufl.edu\)](#). As stated previously, you are required to have a functioning webcam, microphone, and speakers for proctored exams. See the minimum technical requirements at [honorlock.com/support](https://honorlock.com/support). Ensure your computer system meets their minimum system requirements. You are also required to have a handheld mirror/reflective surface for proctoring.

To alleviate the stress of exams, we have incorporated an “**average/replace**” policy (the lowest of the three progress exams will be replaced by the average of the three progress exams). This “average/replace” policy will help to minimize the impact of a single poor performance but it will not completely disappear. For example, if a student has: Exam 1 score of 190/200, Exam 2 score of 150/200 and Exam 3 score of 180/200, then their average/replace score will be 173/200 and it will replace the original Exam 2 score. **Any and all exam grade disputes must be performed within one week of the scheduled exam date.** 5 points will be deducted from your score if you neglect to sign the Honor Pledge question at the end of every exam.

**EXAM ABSENCES:** Absences will be handled in accordance with official UF academic regulations. For more information, see <https://catalog.ufl.edu/UGRD/academic-regulations/>. See below for further clarification for two different types of situations.

(1) Conflicts with other events: Acceptable reasons to miss a scheduled exam include conflicting evening exams in courses with higher course numbers, religious holidays, military obligations, special curricular requirements (e.g., attending professional conferences), or participation in official UF-sanctioned activities such as athletic competitions, etc. For more information on such absences see the official UF Policy at <https://catalog.ufl.edu/UGRD/academic-regulations/attendance-policies/#absencestext> ). If you must be absent for an exam due to a documented and approved conflict known in advance, you must e-mail your instructor the documentation at least one week prior to the scheduled exam and an early conflict exam will be scheduled for you.

(2) Missing an exam due to an emergency or sudden illness: If you are absent for an exam due to an unpredicted documented medical reason or family emergency, you must contact the instructor as soon as possible, and you may be asked to have your excuse verified by the Dean of Students Office (DSO). Your instructor will follow UF academic regulations in evaluating the notification and/or documentation received by you or by the DSO on your behalf. Once your instructor is satisfied with the validity of your exam absence a make-up exam will be scheduled after a reasonable amount of time, i.e., before the end of the semester. If your documentation is deemed insufficient to excuse your absence you will receive a zero on the missed exam.

**HONOR CODE:** UF students are bound by The Honor Pledge which states, “We, the members of the University of Florida community, pledge to hold ourselves and our peers to the highest standards of honor and integrity by abiding by the Honor Code. On all work submitted for credit by students at the University of Florida, the following pledge is either required or implied: “On my honor, I have neither given nor received unauthorized aid in doing this assignment.” The Honor Code (<https://sccr.dso.ufl.edu/policies/student-honor-code-student-conduct-code/>) specifies a number of behaviors that are in violation of this code and the possible sanctions. Furthermore, you are obligated to report any condition that facilitates academic misconduct to appropriate personnel. If you have any questions or concerns, please consult with the instructor or TAs in this class. **5 points will be deducted from your score if you neglect to sign the Honor Pledge question at the end of every exam. You will receive a 0 for the exam and no average/replace, if cheating has been detected.**

**DISABILITIES:** Students with disabilities who experience learning barriers and would like to request academic accommodations should connect with the disability Resource Center by visiting [disability.ufl.edu/students/get-started](https://disability.ufl.edu/students/get-started). It is important for students to share their accommodation letter with their instructor and discuss their access needs, as early as possible in the semester. Once registered, students will receive an accommodation letter which must be presented to the instructor when requesting accommodation (use canvas email). The student is responsible for scheduling the exam dates with the DRC. Students with disabilities should follow this procedure as early as possible. The DRC has 4 business day policy to submit Accommodated Testing Requests (ATRs).

**UF MULTICULTURAL & DIVERSITY AFFAIRS:** Department within the Division of Student Affairs. Multicultural and Diversity Affairs (MCDA) celebrates and empowers diverse communities and advocates for an inclusive campus for all students across identities. MCDA is located on the second level in the student union. <https://multicultural.ufl.edu/>

**U MATTER, WE CARE:** Your well-being is important to the University of Florida. The U Matter, We Care initiative is committed to creating a culture of care on our campus by encouraging members of our community to look out for one another and to reach out for help if a member of our community is in need. If you or a friend is in distress, please contact [umatter@ufl.edu](mailto:umatter@ufl.edu) so that the U Matter, We Care Team can reach out to the student in distress. A nighttime and weekend crisis counselor is available by phone at 352-392-1575. The U Matter, We Care Team can help connect students to the many other helping resources available including, but not limited to, Victim Advocates, Housing staff, and the Counseling and Wellness Center. Please remember that asking for help is a sign of strength. In case of emergency, call 9-1-1.

**COUNSELING AND WELLNESS CENTER:** Visit [counseling.ufl.edu/](https://counseling.ufl.edu/) or call 352-392-1575 for information on crisis services as well as non-crisis services.

**UF TEACHING CENTER (CLAS):** Broward Hall, 352-392-2010 or to make an appointment 352-392-6420. General study skills and tutoring. [teachingcenter.ufl.edu/](https://teachingcenter.ufl.edu/). Check: <https://teachingcenter.ufl.edu/>

**INCLUSIVE LEARNING ENVIRONMENT:** We embrace the University of Florida's Non-Discrimination Policy, which reads, "The University shall actively promote equal opportunity policies and practices conforming to laws against discrimination. The University is committed to non-discrimination with respect to race, creed, color, religion, age, disability, sex, sexual orientation, gender identity and expression, marital status, national origin, political opinion or affiliations, genetic information and veteran status as protected under the Vietnam Era Veterans' Readjustment Assistance Act." We are committed to fostering an open and inclusive classroom and laboratory environment in our College, where every student, guest instructor and contributor feels valued. If you have questions or concerns about your rights and responsibilities for inclusive learning environment, please see your instructor or refer to the Office on Multicultural & Diversity Affairs Website: <http://www.multicultural.ufl.edu/>

**EVALUATIONS:** Students are expected to provide professional and respectful feedback on the quality of instruction in this course by completing course evaluations online via GatorEvals. Guidance on how to give feedback in a professional and respectful manner is available at <https://gatorevals.aa.ufl.edu/students/>. Students will be notified when the evaluation period opens, and can complete evaluations through the email they receive from GatorEvals, in their Canvas course menu under GatorEvals, or via <https://ufl.bluera.com/ufl/>. Summaries of course evaluation results are available to students at <https://gatorevals.aa.ufl.edu/public-results/>.

## GENERAL EDUCATION PROGRAM OBJECTIVES:

Physical science courses provide instruction in the basic concepts, theories and terms of the scientific method in the context of the physical sciences. Courses focus on major scientific developments and their impacts on society, science and the environment, and the relevant processes that govern physical systems. Students will formulate empirically-testable hypotheses derived from the study of physical processes, apply logical reasoning skills through scientific criticism and argument, and apply techniques of discovery and critical thinking to evaluate outcomes of experiments. These objectives will be accomplished through participation in the course lectures and discussion sections, and individual work done on homework assignments and assessments. This course satisfies the General Education requirement in the Physical Sciences. A minimum grade of C is required for general education credit.

**GENERAL EDUCATION STUDENT LEARNING OUTCOMES:** The following learning outcomes will be assessed through online assessments and examinations.

Area	Institutional Definition	Institutional SLO
<b>CONTENT</b>	Content is knowledge of the concepts, principles, terminology and methodologies used within the discipline.	Students demonstrate competence in the terminology, concepts, methodologies and theories used within the discipline.
<b>COMMUNICATION</b>	Communication is the development and expression of ideas in written and oral forms.	Students communicate knowledge, ideas, and reasoning clearly and effectively in written or oral forms appropriate to the discipline.
<b>CRITICAL THINKING</b>	Critical thinking is characterized by the comprehensive analysis of issues, ideas, and evidence before accepting or formulating an opinion or conclusion.	Students analyze information carefully and logically from multiple perspectives, using discipline specific methods, and develop reasoned solutions to problems.

## SPECIFIC GOALS OF CHM2046:

You will be required to analyze scientific concepts and think critically. This means being able to answer both quantitative (mathematical) and conceptual (qualitative) multiple choice problems in a limited period of time. Additionally, you will have to write or orally communicate during your discussion periods. We will also demonstrate how these topics can be applied to the scientific method and how observation and experimentation leads us to the development of scientific theories. To achieve this, students will be introduced to the following concepts from the textbook. You will review the importance of chemistry in our everyday lives. You will be required to utilize the methods of science as a logical means of problem solving through critical thinking. This means you must analyze information carefully and logically from multiple perspectives, using discipline specific methods, and develop reasoned solutions to problems. To ensure your competency in these concepts you will be required to complete online homework assignments and take quizzes and exams that require critical thinking, analysis of problems and drawing conclusions.

**Critical Thinking:** Critical thinking skills are essential in the general chemistry course. There are six criteria by which we promote critical thinking: 1. Information acquisition: Identifying and differentiating questions, problems and arguments. 2. Application: Assessing the suitability of various methods of reasoning and confirmation when approaching a problem. Students are taught to develop hypotheses and to find support and limitations associated with their hypotheses. 3. Analysis: Identifying and analyzing stated and unstated assumption and using logical reasoning to evaluate different viewpoints. 4. Synthesis: Students are encouraged to formulate questions and problems, construct arguments to address such questions and be able to effectively communicate conclusions. 5. Communication: In discussion of alternative points of view, students will be encouraged to criticize or defend their arguments with the use of logical reasoning and evidence. 6. Evaluation: Assessing the quality of evidence and reasoning to draw reasonable conclusions.

**Mathematics:** It is crucial in the general chemistry course to be competent in mathematics. Listed are the criteria by which we promote understanding and application of math: 1. Information acquisition: Students learn to select data that is pertinent to solving a problem. 2. Application: Use of algebraic, geometric and statistical reasoning to solve problems. 3. Analysis: Interpret and draw conclusions from formulas, graphs

and tables. 4. Synthesis: To associate patterns and observations to more abstract principles and to consider specific applications of such principles. 5. Communication: Communicating information symbolically, graphically, numerically and verbally. 6. Evaluation: Estimate and verify solutions to mathematical problems to determine reasonableness, compare alternatives and select optimal results and understand the limitations of mathematical and statistical methods.

**DISCLAIMER:** This syllabus represents my current plans and objectives. If those need to change as the semester progresses, then the changes will be communicated to the class clearly.