CHM 6680 **Spring**, 2021

Characterization of Paramagnetic Molecules (Section 22287 3 credits)

Prof. George Christou CLB 408A Email: christou@chem.ufl.edu

Class Times: M, W, F: Period 4 Room: CCB 221 Office Hours: W 2.00-4.00, and

by appointment

Description: Metal-containing paramagnetic molecules are commonly encountered in chemistry and biochemistry, in fields ranging from inorganic chemistry to the active sites of metalloproteins and metalloenzymes. In all these areas, the presence of the unpaired electron(s) has a major impact on the observed physical and spectroscopic properties. Many chemical and biochemical systems contain mononuclear metal ions, whereas others contain polynuclear sites comprising two or more interacting metal ions. The course will therefore cover the qualitative and quantitative tools and theories needed to understand the magnetic and spectroscopic properties of molecular metal compounds (both mononuclear and multinuclear) possessing unpaired electrons.

Outline:

Basic concepts and definitions in magnetism

Magnetization and magnetic susceptibility

Spin and orbital angular momentum

The van Vleck equation

Transition metals vs lanthanides

Zero-field splitting in mononuclear metal compounds High-spin/low-spin equilibrium, and spin crossover

Polynuclear compounds and intramolecular exchange interactions Mechanisms of antiferromagnetic and ferromagnetic interactions

Spin frustration in polynuclear compounds

Single-molecule magnets (molecular nanomagnets)

NMR of paramagnetic molecules

EPR of mononuclear paramagnetic metal complexes (time permitting)

Grade:

The course grade will be determined by two exams during the semester (30% each) and a Final Exam (40%).

Reference Texts (no required text):

- 1. Molecular Magnetism, O. Kahn, VCH Publishers, Mannheim, 1993
- 2. C. J. O'Connor, Prog. Inorg. Chem. 1982, 29, 203
- 3. Physical Methods in Chemistry, R. S. Drago, 1st or 2nd Edition
- 4. Magnetochemistry, R. L. Carlin, Springer-Verlag, Berlin, 1986
- 5. Magnetism and Transition Metal Complexes, F.E. Mabbs and D.J. Machin, Dover Publications, New York, 2008
- 1. Other Information: Honor code statement: see https://sccr.dso.ufl.edu/policies/student-honor-code-student-conduct-code/
- 2. Students with Disabilities: see https://disability.ufl.edu/
- 3. Counseling and Health Care: see https://counseling.ufl.edu/