

Chemical Bond and Spectra

Course Information

Course number: CHM 6470

Course title: Chemical Bond and Spectra I

Credits: 3

Where: Through Zoom. Links will be distributed through canvas.

When: Tuesdays and Thursdays; Period 7-8 (1:55 pm – 3:50pm)

No class on Nov 26th (Thanksgiving)

List of Recommended Textbooks:

Molecular Quantum Mechanics by P.W. Atkins and R.S. Friedman

Quantum Chemistry by Ira N. Levine

Elements of Quantum Mechanics by M. Fayer

Group Theory and Quantum Mechanics by Michael Tinkham

Online material: Available through canvas (<https://elearning.ufl.edu/>)

It will contain the syllabus, homework and lecture notes. It will also serve as a mode of email communication (you can configure it to redirect the emails).

Instructor Information

Alberto Perez

Assistant Professor

Office: Room 240F Leigh Hall.

Office hours: Friday 10am - 12 pm.

Contact e-mail: perez@chem.ufl.edu

Remote learning

Attendance: Attendance to lectures is expected. We will have synchronous lectures over zoom, lectures will be recorded and made available after class. The 2-period class will combine a traditional lecture with discussion of problems as time allows. In addition to the 4 hrs class a week, ~8-10/week of reading, homework and general study are required.

Students who participate with their camera engaged or utilize a profile image are agreeing to have their video or image recorded. If you are unwilling to consent to have your profile or video image recorded, be sure to keep your camera off and do not use a profile image. Likewise, students who un-mute during class and participate orally are agreeing to have their

voices recorded. If you are not willing to consent to have your voice recorded during class, you will need to keep your mute button activated and communicate exclusively using the "chat" feature, which allows students to type questions and comments live. The chat will not be recorded or shared. As in all courses, unauthorized recording and unauthorized sharing of recorded materials is prohibited.

Homework and Grading

Homework: Homework due date is posted on the class calendar. Late homework (if it is turned in on the same day, but after the deadline) will have a 20% deduction on the grade. The day after, the solutions will be posted, and no more homework will be accepted for grading, although you are still responsible to finish it to be ready for the exams. Each homework problem has to show the full derivation, using SI units. No points will be given for a final result without justification.

Python notebooks: When possible, we'll try to use python notebooks for homework and graphics. This provides a great environment for paper publication ready figures as well as a great environment to handle data.

Exams: There will be 2 progress exams. Conflicts with these exam dates (travel to conferences) must be resolved with the instructor no later than 5 days prior to the exam date. Emergency situations (sickness, death in the family, etc) have to be communicated to the instructor within 48 hrs of the exam and will be considered at the discretion of the instructor.

Final Projects: Consist of two parts. More details will be forthcoming during the year.

1. A python script or python notebook for running a Huckel or Extended Huckel calculation. This can be done in groups.
2. An end of year TED-like formatted talk to make QM topics more accessible to the general public.

Course Grading: The grade will be determined by homework (25%), progress tests (35%), final project (30%) and in-class participation (10%).

There is NO FINAL in this class.

Course Evaluations: Students are expected to provide feedback on the quality of instruction in this course by completing online evaluations at <https://evaluations.ufl.edu>. Evaluations are typically open during the last two or three weeks of the semester, but students will be given specific times when they are open. Summary results of these assessments are available to students at <https://evaluations.ufl.edu/results/>.

HONOR CODE The student honor code can be found at <http://www.registrar.ufl.edu/catalog/policies/students.html>

The students, instructor and TAs are honor bound to comply with the Honors Pledge:

We, the members of the University of Florida community, pledge to hold ourselves and our peers to the highest standards of honesty and integrity.

On all work submitted for credit by students at the university, the following pledge is either required or implied: On my honor, I have neither given nor received unauthorized aid in doing this assignment.

Cheating on an exam will result in a grade of zero. Although homework is expected to be worked “in group”, the submitted homework solution must be your individual work. If any homework or quiz assignment is suspect, a grade of zero will also be given for that assignment.

Information on current UF grading policies for assigning grade points is available at <https://catalog.ufl.edu/ugrad/current/regulations/info/grades.aspx>

Students with disabilities: Students requiring special accommodations need to register at the Dean of Student Offices and bring the documentation to the instructor.

Counseling services are available at <http://www.counsel.ufl.edu> , or call (352)-392-1575 during regular service hours (8am-5pm). For other hours or weekends call the Alachua County Crisis Center (264-6789). Students may also call the clinician on-call at Student Mental Health for phone callback and consultation at (352)-392-1171.

Course Objectives

By the end of this course students should be able to:

1. Understand the need of a mathematical formalism to describe the behavior of atoms and molecules: quantum chemistry
2. Describe the meaning of the wavefunction as it relates to atoms and molecules
3. Apply their knowledge on quantum chemistry and wavefunctions to solve the Schroedinger equation for simple cases (e.g. harmonic oscillator)
4. Explain and apply approximate methods (variation method and perturbation theory) to solve more complex cases (e.g. polyatomic molecules)
5. Relate the mathematical knowledge and applications to experimental observables (e.g. spectroscopic signals)

Course itinerary:

- **Working with python notebooks**
- **Mathematical review:** Differential equations, complex numbers, matrices, waves, eigenvalues/eigenvectors, operators, Dirac Notation.
- **Quantum behavior**

TENTATIVE Schedule of Classes

CHM 6470 FALL 2020

September	
Tuesday	Thursday
1	3
Introduction to the course. Introduction to Python	Mathematical Review: Differential equations. Waves
8	10
Superposition of waves. Standing waves. Matrices	Matrices, eigenvalues and eigenvectors (Levine Molec Spec. Chp 2)

September	
Tuesday	Thursday
15	17
Operators (Chp 3 Levine Quantum CHEM)	Dirac Notation. Operators, Commutators, Stern-Gerlach Experiment (app) Quantum Behavior/Wave Particle duality (ppt)
22	24
Uncertainty Principle. Born Interpretation. Free particles, Localization of wavepackets.	Schrodinger and Heisenberg Representations Particle in a box
29	
Particle in a box (infinite, finite)	

October	
	1
	1-D 2-D finite potentials
6	8
Harmonic Oscillator (ladder operators)	Harmonic oscillator functions, graphics, calculation of observables . Ang. Mometum
13	15
Angular Momentum (with algebra of Operators) Particle in a RING	Rigid Rotors. Schroedinger equation, separation of variables, Functions, degeneracy, Solutions (gral and in particular)
20	22
The H atom (SOLVING THE DIFFERENTIAL EQUATION)	H-atom: THE SOLUTIONS SPIN

27	29
2-state coupled system Coupling of angular momenta	Approximation Methods: Variational Theorem

November

Tuesday	Thursday
3	5
Approximation Methods: Perturbation Theory	Approximation Methods: Perturbation Theory
10	12
Time dependent Perturbation Theory: Spectroscopy	Hartree-Fock Self Consistent Field
17	19
Ab-initio methods and DFT	Diatomic Molecules
24	26
Diatomic Molecules/Semiempirical (Huckel)	Thanksgiving, No class

December

Tuesday	Thursday
1	3
GROUP THEORY	GROUP THEORY (spectroscopy)
8	10
Project presentation	Project presentation