



CHM 4411: PHYSICAL CHEMISTRY I

Equilibrium and Change

Spring 2013, Section 7395, Period 3, Room TUR 2334, MTWF

Instructor: S. O. Colgate, 314 CLB 392-2155 colgate@chem.ufl.edu

Office Hours: T Per 4 and by appointment.

TAs: tba

Text: Physical Chemistry P. Atkins and J. de Paula, 9th Ed, Freeman & Co, NY

Note- This text is available in three versions

1. Hard cover (complete book, Chapters 1 – 20)
2. Paperback Split Version (Vol. I, Chapters 1-6 and 20 -23 and Vol. II, Chapters 7 – 19). **Only Vol. I is needed for CHM 44112**
3. eBook online at www.whfreeman.com/pchem

SYLLABUS

CHM 4411 is the first in a two-term sequence, CHM 4411/4412 in undergraduate physical chemistry. This course includes the study of **THERMODYNAMICS**, the **KINETIC MOLECULAR THEORY**, and **REACTION KINETICS**. The material is presented in the *Fundamentals, Chapters 1 – 6*, and *Chapters 20 – 23* of the textbook. Acquiring more than a superficial understanding of all of these subjects in a single term is improbable, and it is necessary to select what material will be well covered and what will be lightly covered or omitted. While it is important to have an acquaintance with this broad subject matter, even though it is impossible to study and thoroughly understand all of it in such a short period, it is equally important for university students to learn something of the deeper meaning of science, its rules, its strengths, its limitations. The instructor's choice for dealing with this problem is this: Weekly reading assignments are made to assure that, over the course of the term, each student

- 1- reads all of this material, as presented in the textbook

- 2- gains sufficient understanding to answer questions and work select problems at the ends of chapters, and
- 3- submits for grading solutions to a set of specific homework problems.

Simultaneously the in-class lectures will have a narrower focus. They begin with a look at the philosophical principles of physical science then explore how the application of those principles can lead and has led to a growing understanding of the nature and behavior of matter and energy.

This classroom process will often probe more deeply into the subset of lecture topics than is covered in traditional textbooks, and ***students will be expected to demonstrate understanding of this lecture material on the exams***. Most exam questions will relate specifically to material presented in class, whether or not that material is included in the reading assignments or represented in the homework assignments. The only way for the student to be sure what might be covered in the exams is to attend class, where the emphasis of topics will be evident. Each student should understand that ***the content of the course is essentially what is presented in class and included in assignments***. The textbook will be helpful, but it is no substitute for understanding the lectures. Supplementary material will often be posted on the class web site to assist with understanding and learning the lecture material. As a minimum, each student should complete each reading assignment, maintain a well-organized book of lecture notes, and work several problems each week. Homework will be assigned weekly. Some problems will be collected for grading. Work that is turned in for grading must be your own. You may use any resource to learn the material, but what you write down ***MUST*** reflect your own understanding. Do not copy one another's work! Doing so is a violation of the UF Honor Code and of basic ethics. Students who cannot resist the temptation to cheat disgrace not only themselves, but the University and all its students, especially their classmates. There will be four in-class exams during the term. Material covered in the exams will be ***principally*** based on that presented and assigned during that exam period. ***The emphasis will be on the material presented in the lectures***. There is no final exam. A numerical grade based on 1000 maximum points will be computed according to the following schedule:

Homework	25%
Five Exams @ 15% each	75%

The final letter grade will be determined from the net total score, 0 – 1000. The instructor cannot state exactly what the point requirements will be for the various letter grades, but you may judge yourself approximately by this schedule:

A	80%
B+	78%
B	70%
C+	68%
C	60%
D+	58%
D	50%

Class attendance: Refer to the official UF policy at <https://catalog.ufl.edu/ugrad/current/regulations/info/attendance.aspx>

All effort should be made to attend all class meetings and take all exams at the scheduled times. Makeup exams for excusable absences will be arranged with the instructor.

Honor Code. Each student is expected to be familiar with the UF Student Honor Code, the provisions of which may be found at these web sites:

<http://regulations.ufl.edu/wp-content/uploads/2012/09/4040.pdf>

<http://regulations.ufl.edu/wp-content/uploads/2012/09/4041.pdf>

Keep on mind that on all work submitted for credit by students at the University of Florida, the following pledge is either required or implied:

"On my honor, I have neither given nor received unauthorized aid in doing this assignment."

Counseling- Students facing difficulties completing the course or who are in need of counseling or urgent help should call the on-campus Counseling and Wellness Center (352-392-1575; <http://www.counseling.ufl.edu/cwc/>).

Special Accommodations Students requesting classroom accommodation must first register with the Dean of Students Office. The Dean of Students Office will provide documentation to the student who must then provide this documentation to the Instructor when requesting accommodation

Important Notice

We will use the web-based software, Sakai, for communication and record keeping. Registered students may log on to this site at <http://lss.at.ufl.edu/> using your Gatorlink ID and password. Look for announcements, assignments, and messages posted there. Keep in touch with one another, check your grades, and post your comments.