CHM 2046 – General Chemistry II – Fall 2011

Instructor:	Professor Adam S. Veige, CLB 412B <u>veige@chem.ufl.edu</u> (email is recommended for contact) Office Hours: M W F period 3
Sections:	0745, 0747, 0748, 0749, 0750, 0751, 0752, 0754, 1900
Lectures:	M W F Periods 4 (10:40 – 11:30 AM) in CLB C130. Classes begin August 22 nd , 2011 Concurrent enrollment in CHM 2046L, MAC 2311, or MAC 2312 is suggested.
Discussions:	Tuesday classes corresponding to the section numbers above meet with a Teaching Assistant (TA) and will start September 6 th .
Teaching Assistants	: Soumya Sarkar (<u>soumya83@ufl.edu</u> , CLB 419), Amrita B. Mullick (<u>amullick@ufl.edu</u> , CLB 417), M Tariq Jan (<u>tariq1@ufl.edu</u> , CLB 417). TA office hours are held in the Chemistry Learning Center (CLC) in Flint Hall 257 (schedules posted there) and will start August 29th .
Text:	Principles of General Chemistry, by Martin S. Silberberg, 2nd Ed, McGrawHill
Description:	CHM 2046 is the second semester of the CHM 2045 – 2046 General Chemistry I and II sequence. Topics include, chemical equilibrium, acid-base and ionic equilibrium, thermodynamics, electro-chemistry, and an introduction to transition metal coordination compounds.
Attendance:	While attendance will not be taken, it is highly recommended that you attend every lecture and discussion period.
Homework:	Problems found at the end of chapters are strongly suggested. While they will not be graded, their purpose is to reinforce concepts discussed in lecture.
Quizzes:	Four announced quizzes will be given during the Tuesday discussion periods found on the schedule. Each quiz will be worth 50 points. The lowest quiz score will be dropped. You must take quizzes with your respective TA only and during the discussion period in which you are registered.
Exams:	Three mid-semester exams will be administered at night (8:15-10:10 pm) during the course of the semester on T Sep 20, R Oct 27, and W Nov 30 . One of the three mid-semester exams will be dropped, but the final exam is mandatory and may not be dropped. The final exam is cumulative. The final exam is scheduled on M Dec 12 and this date is non-negotiable. Students are required to present their gator1 card during exam.
Make-up Policy:	There will be NO make-up exams or quizzes. A missed exam or quiz will be the one dropped.

Calculators:	Only non-programmable calculators are permitted.	
Cell Phones:	Cell phones must be turned OFF during lectures, discussions, and exams.	
Grading:	The final letter grade will be based on a total of 800 points. Point ranges are fixed and will NOT be curved.	
	Exam 1, 2, and 3 (1 drop)400Quizzes 1-4 (1 drop)150Final Exam250Total800	
	800-700 A, 699-660 A–, 659-629 B+, 628-598 B, 597-567 B–, 566-536 C+, 535-505 C, 504-474 C–, 473-443 D+, 442-412 D, 411-382 D–, 381-0 E	
Sakai Postings:	Suggested problems and homework solutions, exam and quiz scores, exam and quiz inswers and solutions will be posted. Each student should check their Sakai account requently since exam rooms will also be posted in Sakai. Do not email the instructor and TAs within Sakai. Use the email listed above on this syllabus. All registered tudents will automatically have access to Sakai using their Gatorlink ID.	
Course Topics:	 Chapter 17 Equilibrium: The extent of Chemical reactions Chapter 18 Acid-Base Equilibria Chapter 19 Ionic Equillibria in Aqueous Systems Chapter 20 Thermodynamics: Entropy, Free Energy, and the Direction of Chemical Reactions Chapter 21 Electrochemistry: Chemical Change and Electrical Work Chapter 22 The Transition Elements and their Coordination Compounds Special Topics (if time permits) 	
Key Dates:	Quiz:T Sep13, T Oct 11, T Nov15, T Dec6Exam:T Sep 20, R Oct 27, WNov30 M Dec12 (Final)Class Begin:Aug 22^{nd} , Last Class: Dec 7^{th} No Class:M Sep5, F Nov4, FNov11, FNov25	
*CUMULATIVE FINAL EXAM MONDAY, DECEMBER 12 3:00-5:00 pm		
Other Information:		
	Honor Code: http://www.chem.ufl.edu/~itl/honor.html Disabilities: http://www.chem.ufl.edu/~itl/disabilities.html Dounseling: http://www.chem.ufl.edu/~itl/disabilities.html	

Disclaimer: The above course information is tentative and subject to change. The instructor reserves the right to make corrections, additions, and/or deletions as the semester progresses. Syllabus corrections will be announced as they occur.