Professor	Jon D. Stewart
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	Phone: 352.846.0743
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Lectures	Monday, Wednesday & Friday, 1 st (8:00 – 9:15 a.m.), 50 Flint Hall
Office hours	Monday, 9:30 – 10:30 a.m. (Dr. Stewart, 102 Leigh Hall)
	Tuesday, 3:30 – 4:30 p.m. (Dr. Stewart, 102 Leigh Hall)
	Thursday, 8:30 – 9:30 a.m. (Dr. Stewart, 102 Leigh Hall)
	In addition to Dr. Stewart's office hours, teaching assistants are available in the Organic Chemistry Learning Center (205 Hernandez Hall) from approximately 10:00 a.m. – 3:00 p.m. Monday – Thursday and 10:00 a.m. – 12:00 p.m. on Friday. The final schedule will be available on the eLearning site. Please take advantage of all the help that is available to give you the best chance of succeeding in this class.
	We are fortunate to have two teaching assistants (Cristina Iturrey , iturreyc@ufl.edu and Jarrad Pazda , jarradpazda@ufl.edu) assigned full-time to this class. They will have office hours to provide extra practice and help with any problems you may have.
Textbooks	Brown, Foote, Iverson and Anslyn, <i>Organic Chemistry</i> , 6 th or 7 th Edition, Cengage (required).
	Brown, Foote, Iverson and Anslyn, <i>Student Study Guide and Solutions Manual for Organic Chemistry</i> , 6 th or 7 th Edition (recommended).
eLearning (Canvas)	<u>http://lss.at.ufl.edu</u> . Log in with your GatorLink ID and password. This site will be updated regularly with announcements, practice materials, exam scores, recommended homework problems and other information.
Prerequisites	CHM 2210 or the equivalent with a minimum grade of C (2.0).
Course Objectives	This is the second course in the two-semester organic chemistry sequence. This class will continue coverage of fundamental concepts of organic chemistry, including the structure and reactivity of organometallic compounds, aldehydes, ketones, carboxylic acids and their derivatives, enolates, aromatic compounds and amines. Organic spectroscopy and synthesis will also be discussed.

Exams	 Three progress exams and a final exam will be given during regular class times on the following dates: Exam 1 – Wednesday, May 31 Exam 2 – Friday, June 30 Exam 3 – Wednesday, July 26 Final Exam – Wednesday, August 2
	The material covered by each exam can be found on the course eLearning site. Exams will be given only at the scheduled times. Since students can drop their lowest score from exams 1, 2 or 3, no makeup exams will be given for this course. Students who miss an exam due to extreme, unusual circumstances (serious illness requiring a doctor's attention, death in the family, etc.) may request that their final exam score be used to replace the missed progress exam. This option is only available if Dr. Stewart is notified within 24 hours of missing the exam and if proper documentation (doctor's note, funeral program, etc.) is provided. Please note that inadequate preparation because of other academic or extracurricular activities (no matter how worthwhile), is not considered a viable excuse for special consideration.
Grading	Three examinations (100 points each) along with the final examination (100 points), which will be comprehensive, although it will concentrate (approximately 50%) on material presented after the third in-class examination. After each exam, approximate letter grade distributions will be posted so that you will have a feel for your performance relative to others in the class as the semester progresses. The lowest grade from exams 1 - 3 will be dropped before calculating your final grade (you may not drop the final exam score). Your final letter grade will be calculated in two ways:
	1) Points method. After dropping the exam score (from tests 1, 2 or 3) with the lowest number of points, the remaining two scores will be added together with the final exam score and compared to the distribution of total points for the class in order to assign a final letter grade. The class-wide mean of grades assigned by this method will be at the B- / C+ border.
	2) Letter grade method. After dropping the lowest exam letter grade (from tests 1, 2 or 3), the remaining two letter grades will be averaged with that from the final exam by assigning points in the following manner: $A = 4.00$, $A - = 3.67$, $B + = 3.33$, $B = 3.00$, $B - = 2.67$, $C + = 2.33$, $C = 2.00$, $C - = 1.67$, $D + = 1.33$, $D = 1.00$, $D - = 0.67$, $E = 0.00$. The three best values will be averaged, then the following scheme will be used to convert this to the final course grade: 3.85 - 4.00 = A 3.51 - 3.84 = A - 3.18 - 3.50 = B + 2.85 - 3.17 - B

	2.51 - 2.84 = B -
	2.18 - 2.50 = C +
	1.85 - 2.17 = C
	1.51 - 1.84 = C -
	1.17 - 1.50 = D +
	0.84 - 1.16 = D
	0.51 - 0.83 = D -
	<0.51 = E
	For example, if your three best exam letter grades are A, A and A-, your average would be $(4.00 + 4.00 + 3.67) / 3 = 3.89$, which is an A.
	Whichever method (#1 or #2) gives you a higher grade will be used to calculate the letter grade reported to the Registrar.
	Current UF grading policies can be found at <u>https://catalog.ufl.edu/ugrad/current/regulations/info/grades.aspx</u> .
Class Attendance	While attendance is voluntary, the lectures are an essential component of the experience for this class. Readings from the textbook serve as a starting point and the classroom lectures will explain and expand upon this material.
Academic Honesty	We, the members of the University of Florida community, pledge to hold ourselves and our peers to the highest standards of honesty and integrity. On all work submitted for credit by students at the University of Florida, the following pledge is either required or implied: "On my honor, I have neither given nor received unauthorized aid in doing this assignment." <i>All portions of the quizzes and lab reports are to be completed</i> <i>individually</i> .
Students with Disabilities	Students requesting classroom accommodation must first register with the Dean of Students Office. The Dean of Students Office will provide documentation to the student who must then provide this documentation to the Instructor when requesting accommodation.

Tentative lecture schedule

Date(s)	Chapter	Topics
May 8	12	Course introduction; IR spectroscopy
May 10	13	NMR spectroscopy
May 12, 15	15	Organometallic compounds
May 17, 19, 22, 24	16	Aldehydes and ketones
May 26	Exam 1 review	
May 31	Exam 1	Chapters 12, 13, 15 and 16
June 2, 5	17	Carboxylic acids
June 7, 9, 12	18	Carboxylic acid derivatives
June 14, 16, 26	19	Enolates and enamines
June 28	Exam 2 review	
June 30	Exam 2	Chapters 17, 18 and 19
June 30 July 3, 5	Exam 2 20	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions
June 30 July 3, 5 July 7, 10	Exam 2 20 21	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity
June 30 July 3, 5 July 7, 10 July 12, 14, 17	Exam 2 20 21 22	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives
June 30 July 3, 5 July 7, 10 July 12, 14, 17 July 19, 21	Exam 2 20 21 22 23	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives Amines
June 30 July 3, 5 July 7, 10 July 12, 14, 17 July 19, 21 July 24	Exam 2 20 21 22 23 Exam 3 review	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives Amines
June 30 July 3, 5 July 7, 10 July 12, 14, 17 July 19, 21 July 24 July 26	Exam 2 20 21 22 23 Exam 3 review Exam 3	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives Amines Chapters 20, 21 and 22
June 30 July 3, 5 July 7, 10 July 12, 14, 17 July 19, 21 July 24 July 26 July 28	Exam 2 20 21 22 23 Exam 3 review Exam 3 24	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives Amines Chapters 20, 21 and 22 Catalysis
June 30 July 3, 5 July 7, 10 July 12, 14, 17 July 19, 21 July 24 July 28 July 28 July 31	Exam 2 20 21 22 23 Exam 3 review Exam 3 24 Exam review	Chapters 17, 18 and 19 Conjugated systems and pericyclic reactions Benzene and aromaticity Reactions of benzene and its derivatives Amines Chapters 20, 21 and 22 Catalysis